

Atomic Scale Processing Mini-Symposium Room 116 - Session AP2+EM+PS+TF-TuM

Atomic Layer Etching II: Energy-Enhanced Processes

Moderators: Steven M. George, University of Colorado at Boulder, Austin Minnich, California Institute of Technology

11:00am AP2+EM+PS+TF-TuM-13 Atomic Layer Etching of Lithium Niobate for Quantum Photonics, Austin Minnich, California Institute of Technology

INVITED

Lithium niobate (LiNbO₃, LN) is a ferroelectric crystal of interest for integrated photonics owing to its large second-order optical nonlinearity and the ability to impart periodic poling via an external electric field. However, on-chip device performance based on thin-film lithium niobate (TFLN) is presently limited by optical loss arising from corrugations between poled regions and sidewall surface roughness. Atomic layer etching (ALE) could potentially smooth these features and thereby increase photonic performance, but no ALE process has been reported for LN. Here, we report a directional ALE process for x-cut MgO-doped LN using sequential exposures of H₂ and SF₆/Ar plasmas. We observe etch rates up to 1.01 +/- 0.05 nm/cycle with a synergy of 94%. We also demonstrate ALE can be achieved with SF₆/O₂ or Cl₂/BCl₃ plasma exposures in place of the SF₆/Ar plasma step with synergies above 90%. When combined with a wet post-process to remove redeposited compounds, the process yields a 50% decrease in surface roughness. With additional optimization to reduce the quantity of redeposited compounds, these processes could be used to smoothen surfaces of TFLN waveguides etched by physical Ar⁺ milling, thereby increasing the performance of TFLN nanophotonic devices or enabling new integrated photonic capabilities.

11:30am AP2+EM+PS+TF-TuM-15 Tunable Electron Enhanced Etching of β -Ga₂O₃ Using HCl Reactive Background Gas and Positive Sample Voltage, Michael Collings, University of Colorado Boulder; J. Steele, D. Schlom, H. Xing, Cornell University; S. George, University of Colorado Boulder

Crystalline β -Ga₂O₃ is an ultra-wide band gap material with important applications for high power electronics. High precision etching is required for β -Ga₂O₃ device fabrication. Previous thermal atomic layer etching (ALE) attempts to etch β -Ga₂O₃ have not been successful. Plasma etching of β -Ga₂O₃ using Cl-containing gases is difficult for Ångstrom-level etching control and can leave surface damage. In this work, electron-enhanced etching of β -Ga₂O₃ is performed using a HCl reactive background gas (RBG) and positive sample bias. The β -Ga₂O₃ is a -oriented epitaxial film grown by suboxide molecular-beam epitaxy on a single-crystal (0001) Al₂O₃ substrate. The ~100 eV primary electrons from a hollow cathode plasma electron source (HC-PES) are incident on the β -Ga₂O₃ sample. The HC-PES is a chemically robust electron source capable of delivering >200 mA over an area >10 cm². The HCl reactive background gas (RBG) is present at ~1 mTorr. A small positive voltage of <50 V is applied to the sample stage.

The β -Ga₂O₃ film thickness was monitored using in situ spectroscopic ellipsometry during electron exposure. Figure 1 shows that the etching of β -Ga₂O₃ is tunable from 1-50 Å/min by varying the stage voltage from 0 to +40 V, respectively. No etching was monitored from electron exposures without the HCl RBG. Negligible etching was observed without a positive sample stage. The following mechanism can explain these results: (1) The primary electrons at ~100 eV can generate secondary electrons from the substrate. (2) The lower energy secondary electrons can attach to the HCl gas in the reactor. (3) The electron attachment then dissociates HCl into H + Cl⁻ through dissociative electron attachment ionization. (4) The Cl⁻ negative ions are attracted to the sample by the positive sample stage voltage. (5) The incoming Cl⁻ flux leads to an enhancement in etch rate resulting from the formation of volatile chloride species. An illustration of this proposed mechanism is shown in Figure 2.

The β -Ga₂O₃ etching was reproducible and only weakly dependent on the primary electron energy from 100-150 eV. The electron current going to ground through the sample also increased with positive sample voltages as expected if the secondary electrons are pulled back to the sample. Surface morphology of the β -Ga₂O₃ was also investigated after the electron enhanced etching. The RMS roughness decreased after etching. The RMS roughness decreased from 1.88 nm to 1.58 nm after the etching process removed 10 nm at +10 V stage voltage. In contrast, the surface roughness did not change after only electron exposures in the absence of HCl RBG.

11:45am AP2+EM+PS+TF-TuM-16 Bias-Pulsed Atomic Layer Etching, Julian Michaels, University of Illinois at Urbana-Champaign; N. Deegan, Argonne National Laboratory, USA; Y. Tsaturyan, University of Chicago; R. Renzas, University of Nevada Reno; G. Eden, University of Illinois at Urbana-Champaign; D. Awschalom, University of Chicago; J. Heremans, Argonne National Laboratory, USA

Atomic layer etching (ALE) is a binary cyclical process noted for its ability to controllably remove atomic monolayers for nanotechnological device fabrication; however, its relatively slow effective etch rate, often less than a monolayer each minute (tenths of an Angstrom per second), limits its applicability to niche devices that necessitate unmatched precision. If the process were slightly faster, semiconductor, optical, and quantum devices would be regularly implementing ALE because it can reach critical dimensions more consistently while simultaneously offering a smoother post-etch surface, both of which deliver superior device performance regardless of the application.

ALE is slow because the chemical reagent gas(es) is purged in and out of the chamber during each cycle. This is standard practice so that the binary steps, chemical and physical, are fully separated, but purge steps also tend to be the most time-consuming components of any ALE process. Thus, if shortening the cycle duration (speeding up the effective etch rate) is the aim, minimizing purge step duration is a sensible first pursuit.

Bias-pulsed atomic layer etching (BP-ALE) is the execution of the above goal with plasma etching in perhaps the simplest way possible. While traditional plasma ALE often pulses gas flows, plasma DC bias, chamber pressure, substrate temperature, and other parameters, BP-ALE achieves atomic precision by pulsing merely the plasma DC bias, hence "bias-pulsed" indicates that the plasma DC bias is the only parameter that distinguishes the steps, and the purge step duration is absolutely minimized as there is no gas purging whatsoever.

As of now, BP-ALE has been demonstrated in both 4H-SiC and diamond, where the usual ALE cycle duration that often exceeds a minute is superseded by the 6-second cycles of BP-ALE, and smoothing to subangstrom RMS surface roughness is achieved for both after etch treatment. This talk seeks to outline the main differences between BP-ALE in execution and potential applications, explain the material and chemistry characteristics needed for a viable BP-ALE method, and predict material/chemistry systems that are suitable for BP-ALE processing.

12:00pm AP2+EM+PS+TF-TuM-17 Atomic Layer Etching of 2D Transition Metal Dichalcogenides Semiconductors and Its 2D Device Application, Jeongmin Kim, J. Kim, Seoul National University, Republic of Korea

Transition metal dichalcogenides (TMDs), a class of 2D materials, possess a layered structure with individual layers bound by van der Waals forces. TMDs exhibit excellent electrical properties and have potential applications in various fields due to their low surface defect density. Moreover, TMDs possess a unique property where the bandgap varies with the number of layers, leading to changes in electrical characteristics depending on the thickness. Therefore, precise control of TMD thickness is crucial. Conventional etching methods, such as reactive ion etching, suffer from plasma-induced damage or difficulties in precise thickness control, making them unsuitable for accurate thickness control of TMDs. Consequently, atomic layer etching (ALE) utilizing self-limiting reactions has been devised to achieve precise thickness control at the atomic level.

We developed an ALE process for TMDs utilizing the adsorption of SF₆ gas, which exhibits minimal reactivity at room temperature. In a capacitively coupled plasma system, an etching with constant-thickness for each cycle was achieved by adsorbing SF₆ followed by Ar⁺ ion bombardment at an appropriate radio frequency (RF) power. The etch rate was controllable, and no change in the etching rate was observed within a specific RF power range. This was realized by the self-limiting reaction induced by the decomposition of SF₆ under Ar⁺ ion bombardment where further sputtering does not occur once all of the adsorbed SF₆ molecules are consumed. Using our ALE process, TMD-based recessed-channel FETs were fabricated. The electrical characteristics and device performance were evaluated, showing a decrease in off-current with reduced channel thickness and no significant deterioration in device performance. This ALE method enables precise thickness control while maintaining the crystallinity of the channel in TMD-based electronic devices, contributing to device performance enhancement. The ALE technology developed in this study paves the way of the advanced application with the TMD-based electronic devices.

This work was supported by the Korea Research Institute for defense Technology planning and advancement (KRIT) grant funded by Defense Acquisition Program Administration (DAPA) (KRIT-CT-21-034)

Author Index

Bold page numbers indicate presenter

— A —

Awschalom, David: AP2+EM+PS+TF-TuM-16, **1**

— C —

Collings, Michael: AP2+EM+PS+TF-TuM-15, **1**

— D —

Delegan, Nazar: AP2+EM+PS+TF-TuM-16, **1**

— E —

Eden, Gary: AP2+EM+PS+TF-TuM-16, **1**

— G —

George, Steven: AP2+EM+PS+TF-TuM-15, **1**

— H —

Heremans, Joseph: AP2+EM+PS+TF-TuM-16, **1**

— K —

Kim, Jeongmin: AP2+EM+PS+TF-TuM-17, **1**

Kim, Jihyun: AP2+EM+PS+TF-TuM-17, **1**

— M —

Michaels, Julian: AP2+EM+PS+TF-TuM-16, **1**

Minnich, Austin: AP2+EM+PS+TF-TuM-13, **1**

— R —

Renzas, Russ: AP2+EM+PS+TF-TuM-16, **1**

— S —

Schlom, Darrell: AP2+EM+PS+TF-TuM-15, **1**

Steele, Jacob: AP2+EM+PS+TF-TuM-15, **1**

— T —

Tsaturyan, Yeghishe: AP2+EM+PS+TF-TuM-16, **1**

— X —

Xing, Huili (Grace): AP2+EM+PS+TF-TuM-15, **1**