

Plasma Science and Technology Room 315 - Session PS1-FrM

Novel Chemistries and Materials in Plasma Processing

Moderators: Kenji Ishikawa, Nagoya University, Japan, Thorsten Lill, Lam Research Corporation

8:15am PS1-FrM-1 Low Energy Remote Plasma and Non-Plasma Methods for Highly Selective Etching of Advanced Nanodevices, *Mark Kawaguchi*, LAM Research

INVITED

In the pursuit of higher performance semiconductors, device length scales continue to shrink down to nanometer dimensions with emerging chip designs increasingly dependent on 3D structures to maintain scaling. Memory manufacturers have already deployed 3D NAND devices with ongoing development for 3D DRAM devices. Advanced logic manufacturers have already deployed 3D FinFET devices and are beginning to deploy more complex GAA (Gate-All-Around) 3D devices as well. Building 3D structures has increased the need for more sophisticated selective etching. As these selective etches have become critical to the formation of the device, atomic precision material loss is required with low contrast materials. We review exemplary materials systems where insufficient selectivity control necessitates innovative approaches such as remote-plasma radicals with atomic layer passivation & ultra-low residual ions as well as plasma-less low energy thermal chemical etching.

Dummy polysilicon removal requires full removal of the polysilicon with minimal film loss to dielectric films such as silicon oxide and silicon nitride. Remote-plasma radical etch removes the polysilicon, but it is desirable to improve film damage control due to higher energy reactants reaching the surface. The first approach applies atomic layer passivation to modify the top monolayer of an oxide encapsulation layer. Epitaxial film damage is improved by hindering diffusion of halogen radicals through the encapsulating film. The second approach reduces the higher energy reactant flux to the surface. We characterize the intensity of higher energy reactants by measuring the residual ion flux to the wafer surface. By modifying flow geometry above the wafer, the residual ion flux is reduced over 2 orders of magnitude, in the range of nA/cm², and results in improved material loss.

GAA device integration relies on depositing and subsequently removing SiGe layers to form Si-nanowires or nanosheets. Besides the bulk SiGe etch, we highlight strict film loss requirements that require highly selective native oxide removal prior to removing SiGe layers. We describe a thermal chemical etching approach that applies wet etch mechanisms in a vacuum environment. Etch rates are tunable over a wide range from 2-200 Å/min as well as high selectivity to both silicon and silicon nitride materials. In addition, we have applied a Langmuir-Hinshelwood surface kinetics model to extract steady state reactant surface coverage. We find that the surface coverage is sub-monolayer, and over a broad surface coverage range, localized etching characteristics are well-behaved as measured by AFM roughness.

8:45am PS1-FrM-3 A Novel Si Etch Chemistry for CFET Fin Definition, *Yifeng Zhou*, Hiroto Ohtake, Applied Materials, Inc.

CFET channel design requires upper ~75% fin etch profile to be straight and smooth, in addition to the traditional requirements for profile loading, etch depth loading, line edge roughness, and mask selectivity. New etch chemistry and process structure are necessary to meet those challenges as the conventional Cl₂- and SF₆-based fin etch chemistries ran out of steam.

This study introduces a cyclical process featuring a BCl₃/NF₃ based etching step and a CH₄/SO₂ based passivation step that provides straight and smooth upper 85% profile, with minimal profile and depth loading among various pattern pitches.

Profile control: NF₃ is the main etchant and BCl₃ provides effective passivation for Si and SiGe sidewalls. HBr can provide more passivation, via re-deposition of less volatile SiBr_x etch products. NF₃:BCl₃ and NF₃:HBr gas flow ratios are the main knobs for profile control.

Si/SiGe scalloping: Scalloping results from relatively higher lateral etch rate of SiGe compared to that of Si. While SO₂/CH₄ passivation step helps reduce scalloping by forming Ge-S bonds, standalone H₂ trim steps eliminates scalloping by trimming Si faster than SiGe.

Profile loading: iso/dense profile loading roots from sidewall passivation delta between spaces with difference widths. Directional breakthrough step

removes more passivation from more sloped sidewalls to allow for more taper correction in subsequent etch steps.

Depth loading: SO₂/CH₄ passivation step deposits thicker carbon polymer on wider trenches, negating the higher etch rate from higher neutral fluxes.

9:00am PS1-FrM-4 The Effect of Remote Ar Plasma Exposure on the Crystalline Structure of VO₂, *Peter Litwin*, *Neeraj Nepal*, U.S. Naval Research Laboratory; *Marc Currie*, US Naval Research Laboratory; *Andrew Lang*, *Luis Carrillo*, *David Boris*, *Michael Johnson*, *Scott Walton*, *Virginia Wheeler*, U.S. Naval Research Laboratory

The effect of chemically inert plasmas on the surface of thin film materials is of interest because it allows for an additional means to deliver energy to a material beyond increasing the material's temperature. This could be of use in cases where energy needs to be delivered to specific layers or interfaces in a heterostructure without subjecting the entire material stack to higher temperatures or to overcome energetic barriers, such as those associated with material crystallization (or amorphization) or the formation of metastable phases. However, plasmas make for a complex environment due to the various energy contributions from ions, electrons, excited species, and photons. Despite this complexity, an increased understanding of energy transfer by controlling the fluence and energy of ions at the plasma-surface interface opens additional avenues in material engineering. This may be particularly true for plasma-enhanced atomic layer deposition (PEALD) processes where thin films are cyclically exposed to plasmas during deposition and thus could benefit from an increased understanding of the interaction between plasmas and material surfaces.

In this work, we look at the impact of remote Ar plasmas on crystalline VO₂ thin films deposited by ALD as a function of plasma power and substrate bias. Thermal ALD of VO₂ was carried out at 150 °C using TEMAV and O₃ and subsequently crystallized in a dedicated thermal annealing vacuum system following a previously established process¹. We find that the Raman signature of the equilibrium monoclinic phase of VO₂ (m-VO₂) decreases after exposing the sample to a remote Ar plasma beyond a minimum power or substrate bias threshold. Interestingly, this process is reversible, i.e., the signature of m-VO₂ can be recovered by exposing the sample to a subsequent Ar plasma of decreasing power or bias starting below the initial threshold required to reduce the m-VO₂ signature. These results suggest one of two possibilities: either amorphization of the film and subsequent recrystallization or selective stabilization of different VO₂ phases based on the plasma condition used. TEM analysis indicates the plasma-treated material is in fact highly crystalline despite the decreased signature of m-VO₂, which suggests the latter of the two explanations. This presentation will discuss the details of these experiments and, with the aid of in situ plasma diagnostics, attempt to elucidate the plasma parameters responsible for the behavior observed.

¹ J. Phys. Chem. C 2017, 121, 19341–19347

9:15am PS1-FrM-5 Plasma-Surface Interactions of Polyurea Deposited on SiO₂ and SiN_x with HF Plasmas, *Wallis Scholl*¹, Colorado School of Mines; *Mingmei Wang*, *Thorsten Lill*, *Wenyu Zhang*, *Louis Kim*, *Harmeet Singh*, Lam Research Corp; *Sumit Agarwal*, Colorado School of Mines

Fluorocarbon plasmas have been traditionally used to etch SiO₂ and SiN_x during semiconductor processing. Because fluorocarbon radicals inherently lower the etch rate by depositing a polymer on the SiO₂ or SiN_x surface, the industry is transitioning to C-free reactive plasmas, such as HF. As a result, new techniques are needed to enhance surface passivation and etch selectivity during HF plasma etching. During etching of high aspect-ratio (HAR) features, sidewall passivation is needed to preserve the profile. Molecular layer deposition (MLD) is a technique for depositing organic and hybrid organic-inorganic films. We studied the HF plasma interactions after a polyurea film deposited by MLD on top of SiO₂ or SiN_x. We have shown in previous work that MLD polyurea film deposited on top of SiO₂ or SiN_x can protect the underlying material from etching. During HF plasma etching of blanket films on a planar substrate, the polyurea film acted as a sacrificial layer and prevented SiO₂ or SiN_x etch while the polyurea was present. When the polyurea was completely consumed by reaction, the SiO₂ or SiN_x film was etched at its baseline etch rate (see Figure 1). The presence of polyurea also delayed the formation of ammonium fluorosilicate on the SiN_x surface, which is known to form on the surface in the presence of HF, and SiF₄ and NH₃ generated as etch products.

In this work, we also deposited polyurea onto 65:1 aspect ratio holes in SiO₂-SiN_x stacks, which were then exposed to an etching plasma to evaluate sidewall passivation. Scanning electron microscopy images show that the

¹ PSTD Coburn & Winters Student Award Finalist

polyurea film reduced the diameter throughout the entire depth, showing both the conformity of the MLD film as well as successful passivation of the sidewalls. Subsequent etching of these features preserved the critical dimension compared to etching without the polyurea film (see Figure 2). Further, *in situ* infrared spectroscopy was used to monitor polyurea interactions with plasma species to better understand the etch mechanism. We exposed the films to a remote HF plasma and HF gas to mimic the conditions in HAR structures. While the radical species in the remote HF plasma gradually removed the polyurea film, it remained largely intact during exposure to HF gas. Fluorination of the polymer film was observed prior to its removal, suggesting that polyurea is removed as volatile hydrofluorocarbon fragments during HF plasma etching. Additionally, we show that during etching at a higher substrate temperature of 150 C, the presence of polyurea enhanced the SiO₂ etch rate: we attribute this to activation of HF through hydrogen-bonding interactions with the amines in the polyurea film.

9:30am **PS1-FrM-6 Long-Term Influence of Fluorine Containing Plasmas and Gases on Sapphire (Single-Crystal Al₂O₃) and Ni-Based Alloy Surfaces, *Takuya Ishihara*, Nozomi Kida, Hidenobu Tochigi, Keigo Iwamoto, Azbil Corporation, Japan; Kazuo Karahashi, Nagoya University, Japan; Satoshi Hamaguchi, The University of Osaka, Japan**

In semiconductor manufacturing processes such as dry etching or chemical vapor deposition, capacitance manometers are widely used as essential vacuum pressure sensors to monitor and control the pressures of process gases. These gauges must be corrosion-resistant against process gases such as halides and their radicals generated by the plasmas. The diaphragm material of the manometer is especially important because, if its surface is altered by such corrosive gases, the sensor would send imprecise output signals possibly with the zero-point drift or pressure sensitivity shift. The errors are caused by the changes in mechanical properties of the diaphragm arising from the formation of the modified surface layer. For this reason, Ni-based alloys or polycrystalline ceramics of aluminum oxide (Al₂O₃) are typically used as the diaphragm material of capacitance manometers. More recent capacitance manometers employ sapphire (single-crystal α -Al₂O₃) as their diaphragm material, which is of specific interest in this study [1]. Recent studies on the interactions of polycrystalline Al₂O₃ with fluorine-containing plasmas indicated the formation of aluminum fluoride layers on Al₂O₃ exposed to such plasmas [2,3,4,5,6]. We have reported the results of ion beam experiments to understand the surface modification mechanisms of Ni-based alloys and polycrystalline Al₂O₃ film by fluorine-containing plasmas [7,8]. In addition, Ni-based alloy samples were exposed to xenon difluoride (XeF₂) gases for 3 and 6 months and their fluorinated surfaces were analyzed [9]. In this study, we obtained 12 months XeF₂ exposure data of Ni-based alloys, in which fluorine continued to diffuse even after one year, unlike in sapphire. Furthermore, the differences in surface states between sapphire and nickel-based alloys upon exposure to XeF₂ or NF₃ remote plasma will also be reported.

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- [2] Chen Chien-Wei, *et al*, J. Vac. Sci. Technol. A Vol.41 No.1 Page.012602(pp9) (2023)
- [3] Kim Yewon, *et al*, Appl. Surf. Sci. Vol.641 Page. Null (2023)
- [4] Vos Martijn F. J., *et al*, J. Phys. Chem. C Vol.125 No.7 Page.3913-3923 (2021)
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- [6] Fischer Andreas, *et al*, J. Vac. Sci. Technol. A. Vol.38 No.2 Page.022603(pp7) (2020)
- [7] T. Ishihara, *et al*, the AVS 70th International Symposium & Exhibition (2024)
- [8] H. Kang, *et al* Jpn. J. Appl. Phys. Vol. 64 Page: 046001(pp8) (2025)
- [9] T. Ishihara, *et al*, the AVS 71st International Symposium & Exhibition (2025)

9:45am **PS1-FrM-7 Investigations of Oxygen-Containing Plasma Etching of Diamond, *Justin Boles***, University of Houston; *Louis E. S. Hoffenberg*, Jack S. Draney, David. B. Graves, Princeton University; *Vincent M. Donnelly*, University of Houston

Plasma etching of single-crystal diamond is vital for the development of next-generation power electronics and quantum platforms based on nitrogen-vacancy complexes (NV-centers). Despite this, little has been reported on the fundamental etching mechanisms, including the roles of

ions and neutrals. Here we report a detailed study of the etching of single-crystal diamond (100) in O₂/Ar magnetically-confined, inductively-coupled plasmas, using a combination of optical emission spectroscopy, ion flux measurements, etching rate measurements, and simulations. Fluxes of positive ions (O⁺, O₂⁺, Ar⁺) to the (100) diamond surface were determined from ion saturation current measurements and Langmuir probe measurements, while O and O₂ number densities were determined using optical emission actinometry. Optical emission spectroscopy was further employed to measure gas temperature (via N₂ rotational spectroscopy) and identify CO and CO₂ reaction products. Etch rates were measured via masking and optical profilometry as functions of ion energy and %O₂ in Ar. In “pure” (98% or 100%) O₂ plasmas, with an O-to-O₂⁺ flux ratio of about 25:1, the etch rate and product optical emission intensities obeyed a square root of ion energy dependence, with a threshold near 11 eV. The etching yield (C atoms-per-ion) was 7.5 at a mean ion energy (DC self-bias voltage plus plasma potential) of 215 eV. If CO were the only product, and since physical sputtering yields are much lower, O atoms must be playing a dominate role in etching at this energy. As the feed gas ratio was changed from pure O₂ to 20%O₂/Ar at constant pressure (10 mTorr), ICP power (300 W), and ion energy (215 eV), the total positive ion flux remained nearly constant (1-2 X10¹⁶cm⁻²s⁻¹) and the O-to-ion flux ratio actually increased to between 40:1 and 50:1, while the yield decreased monotonically to about 3.5 C-per-ion, still much larger than the sputtering yield of 0.16 C-per-Ar⁺. The yields as a function of O-to-ion flux ratio and %O₂ in Ar compare favorably with molecular dynamics simulations.

This work supported by the U.S. Department of Energy, Office of Science, Office of Fusion Energy Sciences under Award No. DE-SC0024537.

10:00am **PS1-FrM-8 Surface Reactions in Fluorine-Containing Cryo-condensed-Alcohols under Ion Beam Irradiation, *Airah Osonio***, Kazuhiro Karahashi, Nagoya University, Japan; *Chihiro Sakamoto*, Wu Li, *Chihiro Abe*, Toshiyuki Sasaki, Mitsuhiro Omura, KIOXIA, Japan; *Makoto Sekine*, Kenji Ishikawa, Nagoya University, Japan

Cryogenic surface processing offers unique opportunities for controlling the adsorption, transport, and reaction of reactive species with high precision, enabling atomic-scale surface modification based on well-defined reaction mechanisms. In this work, we establish a novel experimental platform that combines cryogenic condensed layers with ion beam irradiation to investigate ion- and radical-assisted surface reactions under highly controlled conditions.

The experimental system integrates a low-energy ion beam apparatus with spectroscopic ellipsometry operating under ultra-high vacuum conditions. Ethanol vapor was deposited onto SiO₂ substrates kept below 180 K, forming cryogenic condensed layers whose evolution was monitored through real-time optical measurements. Dynamic optical modeling of the data enabled extraction of film thickness during deposition, thermal evolution, and ion beam exposure. To explore the behavior of reactive species in the cryogenic medium, fluorine-containing gases were introduced during and after ethanol film formation. Subsequent ion beam irradiation induced modification of the condensed layer and selective removal of the underlying SiO₂ film.

A key advantage of this approach is the independent control of ion flux and reactive neutral species. Unlike conventional plasma etching environments, where ions, radicals, and reaction products interact simultaneously, the ion beam configuration enables systematic separation of these effects, providing a powerful framework for elucidating elementary surface reaction pathways. By varying the ethanol condensation rate, fluorine precursor exposure, and temperature below the condensation threshold, the influence of cryogenic layers on surface reactions can be examined. The results suggest that fluorinated species incorporated within the high-dielectric ethanol condensed layer locally promote etching reactions under ion irradiation. At temperatures below 180 K, the confined cryogenic environment exhibits characteristics analogous to solution-phase chemistry, enabling localized wet-like chemical reactions despite operation under vacuum conditions.

The results demonstrate that cryogenic molecular films influence fluorine-based ion-assisted surface processes. The transient behavior suggests that cryo-condensed ethanol layers may modify the transport, retention, and reaction of reactive species at the substrate interface. The combined ion-beam and ellipsometry platform provides a new framework for investigating adsorption, desorption, and ion-assisted chemical processes at

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solid surfaces, offering insights relevant to next-generation cryogenic processing technologies.

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