

Study on photodeposition process of Pt nanoparticles on TiO₂ photocatalyst by XAFS spectroscopy

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TiO₂ is one of the representative photocatalysts used, for example, to purify water and air. In order to enhance the activity of the TiO₂ photocatalyst, Pt nanoparticles are often deposited on the surface[1]. In this study, we aimed to investigate the photodeposition process of Pt nanoparticles on TiO₂ using Pt L₃-edge XAFS spectroscopy focusing on the bonding states of Pt on TiO₂ surface and local structures around Pt atoms.

Before irradiation, the feature of Pt L₃-edge XANES spectrum of the TiO₂ sample adsorbing the Pt precursor was similar to that of PtO₂. The energy position of the main peak of XANES shifted to lower X-ray energy with the photoirradiation time, indicating a decrease in the ratio of Pt⁴⁺ ions by reduction.

All XANES spectra were reproduced with the linear combination of two XANES spectra of PtO₂ and Pt metal, and the fractions of Pt⁴⁺ and Pt⁰ for all the Pt/TiO₂ samples were directly evaluated. Fig.1 shows irradiation-time dependence of edge jump intensity at 11740 eV and the fraction of Pt⁰. The edge jump intensity corresponds to the deposited Pt concentration.

Before photoirradiation, only 20% of Pt⁴⁺ ions were adsorbed on TiO₂. The Pt⁴⁺ ions increased to 45% by photo-assisted adsorption after 5 minutes, in which most of the adsorption sites of Pt on the TiO₂ surface would be occupied. Subsequently, the adsorbed Pt⁴⁺ ions were gradually reduced during the following 5 to 20 minutes of photoirradiation. In the successive period for 20–30 min, the amount of the adsorbed Pt⁴⁺ ions drastically increased and the reduction of Pt⁴⁺ to Pt⁰ drastically occurred. After that, the fraction of Pt⁰ gradually increased by further photoreduction.

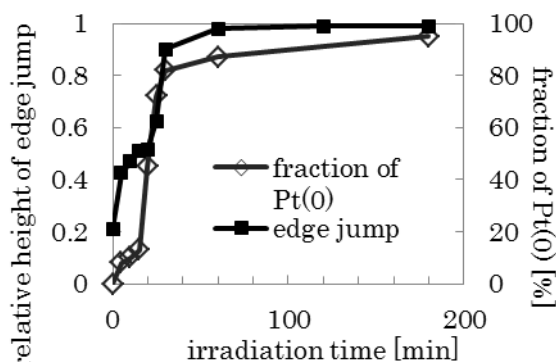


Fig.1 Dependence of relative height of edge jump in Pt L₃-edge XANES and fraction of Pt⁰ on the photoirradiation time.

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